2024 FALL Semester

General Chemistry Lab I (CH102)

Exp.0. Orientation

Department of Chemistry

2024. 9.

Congratulations all of you and Welcome to General Chemistry Laboratory Class !

Today's Topics

- Introduction
- Utilizing Web System for Online Lab Class
- Lab Schedule
- Method of Evaluation / Grading
- Additional Information
- Watching Safety Video in the Laboratory

Introduction

- Course Information
- Course Objective
- Course Requirement

1. Course Information

- General Chemistry Lab I
 - -Number: CH02
 - -Lecture: Exp: Credit = 0:3h:1

2. Course Objective

- 1) To teach basic laboratory techniques
- 2) To introduce to elementary methods of assessing the significance of experimental measurements
- 3) To provide experiences that enable the students to acquire positive attitude toward CHEMSTRY, or SCIENCE

3. Course Requirement

Writing up and submission of lab reports.

- Pre-lab and Post-lab Assignments: Introduction & Questions
- Observations and experimental detail.
- Detailed method of processing the experimental data. (For Quantitative analytical experiments)
- Calculations and Conclusions regarding the accuracy and the precision of experimental results and errors and the inherent errors based on the measurements.

Lab Schedule

- Laboratory Experiments
- Time Table
- Manual
- General Chemistry Website

General Chemistry Lab I (CH102)

Time		Mon	Tue	Wed	Thu	Fri
a.m	9:00~12:00				CH102-A	СН102-Е
					CH102-B	CH102-F
p.m.	13:00~16:00				CH102-C	CH102-G
					CH102-D	СН102-Н
	16:00~19:00					CH102-I
						CH102-J

*Location: GoongNi Laboratory Bldg. (E6-5): Lab 402 & Lab 406

General Chemistry Lab I (CH102)

Experiment # and Topics

- Exp.0. Orientation: Introduction, Lab Schedule, and Method of Evaluation
- Exp.1. A Carbonate Analysis: Molar Volume of Carbon dioxide (Experiment 13)
- Exp.2. Periodic Table and Periodic Law (Experiment 11)
- Exp.3. Quantum Chemical Calculation: The Potential Energy Curve and the Orbitals of H_2^+
- Exp.4. Atomic and Molecular Structure (Experiment Dry Lab3)
- Exp.5. Thermodynamics of the Dissolution of Borax (Experiment 26)
- Exp.6. A Rate Law and Activation Energy (Experiment 24)
- Exp.7. LeChatelier's Principle; Buffers (Experiment 16)
- Exp.8. Galvanic Cells, the Nernst Equation (Experiment 32)
- Exp.9. Alkalinity of a Water Resources (Experiment 20)

CH102 (Week $1 \sim 8$)

	Exp#				
Period	Thursday		Friday		
	Class A, C	Class B, D	Class E, G, I	Class F, H, J	
9/5, 9/6	Expo/ lab402	Expo/ lab4o6	Expo/ lab402	Expo/ lab4o6	
9/12, 9/13	Exp1/ lab402	Exp2/ lab4o6	Exp1/ lab402	Exp2/ lab4o6	
9/19, 9/20	[No class]	[No class]	[No class]	[No class]	
9/26, 9/27	Exp2/ lab406	Exp1/ lab402	Exp2/ lab4o6	Exp1/ lab402	
10/3, 10/4			[No class]	[No class]	
10/10, 10/11	Exp3/ lab4o6	Exp4/ lab402	Exp3/ lab4o6	Exp4/ lab402	
10/17, 10/18	Exp4/ lab402	Exp3/ lab4o6	Exp4/ lab402	Exp3/ lab4o6	
10/21~25	Mid-term Exam period				

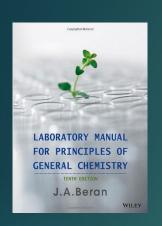
*Location: GoongNi Laboratory Bldg. (E6-5): Lab 402 & Lab 406

CH102 (Week 9 ~ 16)

	Exp#				
Period	Thursday		Friday		
	Class A, C	Class B, D	Class E, G, I	Class F, H, J	
10/31, 11/1	Exp5/ lab402	Exp6/ lab4o6	Exp5/ lab402	Exp6/ lab4o6	
11/7, 11/8	Exp6/ lab4o6	Exp5/ lab402	Exp6/ lab4o6	Exp5/ lab402	
11/14, 11/15	Exp7/ lab402	Exp8/ lab4o6	Exp7/ lab402	Exp8/ lab4o6	
11/21, 11/22	Exp8/ lab4o6	Exp7/ lab402	Exp8/ lab4o6	Exp7/ lab402	
11/28, 11/29	Exp9/ lab402	Exp9/ lab4o6	Exp9/ lab402	Exp9/ lab4o6	
12/5, 12/6	[make-up]	[make-up]	[make-up]	[make-up]	
12/12, 12/13					
12/16 ~ 12/20	Final Exam period				

*Location: GoongNi Laboratory Bldg. (E6-5): Lab 402 & Lab 406

3. Laboratory Manual



Laboratory Manual for Principles of General Chemistry,

10th Edition, J. A. Beran, John Wiley & Sons, Inc.

You can purchase the textbook from online bookstore; https://product.kyobobook.co.kr/detail/Sooooo3110569

http://www.yes24.com/Product/Goods/11074720

Or Sciplus (사이플러스)

http://www.sciplus.co.kr/sp/index.php?inc=books_main&cate=29 &sub=bdetail&no=2514

4. General Chemistry Website www.gencheminkaist.pe.kr

The experimental procedure file with photos will be uploaded continually at a link, [Experimental Procedure] of left frame on the general chemistry website. Please look over the procedure on the manual and the file in advance so that you can complete your work successfully.

Method of Evaluation and Grading

- Grading & Examination Info
- Paint distribution

1. Grading & Examination Info

- ✓ Less than 50% of the total students in each class receive an A grade. Students who score below 50% of the total marks or fail to attend two or more experiments are assigned an F grade.
- ✓ Examination Information: No examination.

2. Point Distribution

The student will be evaluated based on his/her performance of the requirement listed below.

One Experiment = 100 points (9 experiments X 100 points) + (safety training :100 points) = 1000 points in total 1) Laboratory Experiment
The assignment description of an experiment consists of three parts; Pre-lab
Assignment, Attitude, and Lab Report. Points distribution is as follows.

- 1-1) Pre-Lab Assignment (20 points)
 - (a) Introduction (Theory) (5 pts) and Procedure Summary (5 pts)
 - (b) Quiz (10 pts)
- 1-2) Attitude (20 points)
 - (a) lateness, lab safety or cleanup,
 - (b) Concentration or comprehension
- 1-3) Laboratory Reports (60 points)
 - (a) Result
 - (b) Discussion
 - o Summary
 - o Assessing the results
 - o Conclusions
 - (c) Reference
 - (d) Post-lab questions

Part A. Pre-lab Assignment (20 pts)

- Introduction(5 pts) and Procedure Summary (5 pts)(Point Range = 0 ~ 5 points each)
- Question (10 pts) = 5 Problems X 2 points(Point Range = 0 ~ 10 points)

Part B. Attitude (20 points)

- Section: Attendance-Late Coming
- Penalty points 5 pts
- Explanations

When you arrive at the lab before TA calls your name, you get a full score, 5 pts. If you arrive 10 min late after the beginning of the lab, you get 3 pts. When you arrive between 10 min and 20 min, you will get the score, 0 pts. For being 20 minutes late, you can't participate in the lab experiment. (No chance any more.)

- Section: Attendance- Absence without notice
 Penalty points: -100 pts
 Explanations
 - If you are absent 2 times or more without any notices, F credit will be recorded.
 - The lab experiment is ONLY permitted in registered class.
 - Only valid excuses are illness (with a valid doctor's excuse) or unusual circumstance beyond your control (death in family), etc.
 - The student must submit written verification of them for an excused absence.

- Section: Lab Safety & Cleanup
 Penalty points: -5 pts
 Explanations
 - Personal Protective Equipment: Safety glasses, Attire (lab coat) Wear approved eye protection at all times while in the laboratory. The penalty is -5 pts for this.
 - Care of balance: Anyone found to be leaving spilled chemicals in the balance area or bench will lose 5 pts for each violation. Any chemicals spilled on the bench top or the balance pan during transfer are to be cleaned up immediately by the student.
 - Cleanup of Lab Bench: The students are responsible for cleaning up their immediate lab bench area before leaving lab. This means wiping up any spills and disposing of any paper towels from the bench top and sink. Failure to do this result in a deduction of -5 pts.
 - Disposal of chemicals DO NOT discard chemicals down the laboratory sink. The penalty is -5 pts for this

- Section: Concentration & Comprehension
- Penalty points -5 pt
- Explanations

Check lists

- playing with gossiping
- cell phone (touching, ringing, chatting), and so on
- Repeated failure of the same experiment

The penalty is -5 pts for each case.

Part C. Laboratory Report (60 Points)

☐ Result & Discussion (50 pts)

Results 0 Data units

O Graphs or Analysis

O Calculations in a detail

OTables

Discussion (20-40 lines) 0 Summary

O Assessing the results (Analysis)

0 Conclusions

Reference

Post-lab Questions: 10 pts = 5 Problems X 2 points (point range: 0 ~10 pts)

Teaching assistant will choose and assign the number of problems out of *Laboratory Question* of each topic on the lab manual.

2) Online Safety Training Completion

2-1) Route: safety.kaist.ac.kr → Lab Safety Management System → Lab Safety Education → Online Education 2-2) Bachelor's requirement

classification	Hours required	Topics		
Freshmen	-New user education for freshment- 2 hours (within 3 months of admission)	n -New user education: [New E ducation]		
11001111011	- Regular education: 6 hours~	- Regular education: Free cho ice of Chemical topics		
Sophomore ~ Se nior	Regular education: 6 hours ~	- Regular education: Free cho ice of Chemical topics		

^{*} KS Lab (A Laboratory subject to thorough safety inspection): General Chemistry Lab (E6-5)

Utilizing Web System

Online WEBs

- Video & Q & A board

 klmskaist.ac.kr
- Assignment Submission

 www.turnitin.com

 Your TA will let you know information (class ID and enrollment key) to register into your lab class on *Turnitin system via* announcement on the KLMS or via e-mail later. Based on the experimental data, student must submit a file including pre-lab assignment, lab report, and post-lab assignment through the Turnitin system in due date. You should use MS-word to produce your assignments.

Additional Information

- Assignment Submission and Feedback
- Guidelines to check Plagiarism Using the Turnitin Software
- Policy for Late Lab Report
- Counting Attendance in Online Lab Class

1. Assignment Submission and Feedback

- Online Submission: www.turnitin.com
 - Create a user profile. Enter Class ID and enrollment key (provided by TA).
 - Enter your name in Korean or English and email address.
- Submission Due: Within 7 days
- Posting Grade: Within 3 days from the due date
- Qaim Period Within 2 days after the period of grading and feedback

[Example]

Wed	Thu	Fri	Sat	Sun	Mon	Tue
	[Exp1]		Due	: Within	7 days	
	Period for Grading and					
	Feedback of the			Claim	period	
	Assignment					

2. Our Guidelines to check Plagiarism Using the Turnintin Software

Reports having similarity ranges from 24% to 100% or the following common sources will be regarded as plagiarism that results in zero for all reports involved (determined by chief TA and instructor).

Example 1. Text matching

- Overall Similarity index: <u>Should not exceed 24%</u> (24% and below gives the color code-Green, in Turnitin, indicating that it within the acceptable).
- Acceptable number of words in unbroken string (phrase or sentence): <u>Should</u>
 not exceed 15 words

The color of the report icon indicates the similarity score of the report, based on the amount of matching or similar text that was uncovered. The percentage range is 0% to 100%. The possible similarity ranges are:

- Blue No metching text
- Green One word to 24% metching text
- Yellow 25-49% metching text
- Orange 50–74% metching text
- Red -75-100% metching text

TITLE	SIMILARITY
Submission	0%
Submission	6%
Submission	43%
Submission	58%
Submission	80%

If you get an orange 52% similarity percentage, that means that 52% of your paper is exactly the same as other sources found by Turnitin. Even in case of 15% similarity, if the matching text is one continuous block of borrowed material (should not exceed 15 words), it will be considered as plagiarized text of significant concern.

Example 2. Cut/ Copy and Paste material from the Web, textbooks or online manual, data (lifting phrases, sentence and paragraphs of someone's work beyond an acceptable number of words)

Example 3. Copying the work of another student

3. Policy for late lab reports

You should submit your assignment by due date on Turnitin. You are entirely responsible for both upload of the assignment file within due date and confirmation of the successful upload to Turnitin. Várning! Wé will not accept any excuses or compromise in case that you deduct any points below due to your late report submission If you miss last chance, you don't need to submit it.

- ✓ Within 24 hours (last chance): -20 points
- ✓ More than 24 hours: -100 points

Good luck and Best wishes for an error-free and accident-free term.